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fluid-mechanics-5th-edition-by-kundu-cohen-and-dowling/ Chapter 1 Introduction 1.1 A gas at 20°C may be rarefied if it contains less than 10¹² molecules per mm³. If Avogadro’s number is 6.023E23 molecules per mole, what air pressure does this represent? Solution:

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The mass of one molecule of air may be computed as Molecular weight 28.97 mol⁻¹

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Solution 1.1. To get started, first list or determine the volumes involved: V_d = volume of water dumped = 100 cm³, V_c = volume of a sip = 5 cm³, and V_o = volume of water in the oceans = $4\pi R^2 D$, where, R is the radius of the earth, D is the mean depth of the oceans, and f is the oceans' coverage fraction. Here we've ignored the ocean volume occupied by salt and have assumed that the ...

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