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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, H_{00C}-COOH•xH₂₀, by titrating a known mass of the acid with a 0.100 mol dm⁻³ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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[2] (b) Marit arranged the values she found in Table 12 into an energy cycle. C₂H₄(g) + H₂(g) C₂H₆(g) 3 0 2 2CO₂(g) + 3H_{20 - 1 4 1 1 k J m o l - 1 - 2 8 6 k J m o l - 1 - 1 5 6 0 k J m o l - 1} (hydrogenation) (l) Calculate the value for the enthalpy of hydrogenation of ethene from the energy cycle.....

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