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the Periodic Table. You can get the  
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adjusting the number of neutrons  
Adding electrons will change it from a  
+ion to neutral atom, to ion depending  
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Adding electrons you can only add 2 to the first orbital and 8 to the second b)

What particle(s) are found in the center of the atom?

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Explore the Build an Atom simulation.  
a) List two things you observed in the simulation. Proton, neutron and

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electron are different colors The amount of these three changes the element b) What particle(s) are found in the center of the atom? Protons and neutrons 1. Play until you discover which particle(s) determine(s) the name of the element you build.

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Identify an element, its

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Answers - [scheduleit.io](https://www.scheduleit.io)

hydrogen2 and hydrogen3 are both isotopes of hydrogen1 magnesium25 is an isotope of magnesium24 carbon-13 and carbon14 are both isotopes of



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Carbon12 (carbon12 1 is an anion of carbon12, carbon12 +1 is a cation of carbon12) anions are negative ions with more electrons than protons cations are positive ions with fewer electrons than protons (the Gallium examples are also ions) 5 BUILD AN ATOM 6 BUILD AN ATOM

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Sample Learning Goals Use the number of protons, neutrons, and electrons to draw a model of the atom, identify the element, and determine

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the mass and charge. Predict how addition or subtraction of a ...

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☐Build An Atom☐ Simulation ☐ Build an Ion Learning Goals: 1. Draw models that show atomic structure. 2. Use information about the number of protons, neutrons, and electrons to

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Identify an element, its position on the periodic table. 3. Predict how changing the number protons and electrons will change the element and its charge.

[Build An Atom](#) Simulation [Build an Ion](#)

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